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Carbon Compound Study Guide Answer

Covalently bonded compound that contains carbon, excluding carbonates and oxides. Functional Group Portion of a molecule that is active in a chemical reaction and that determines the properties of many organic compounds.

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Biology Section 3-1 Review: Carbon Compounds Flashcards ...

Carbon is a crucial element of our world. Everything that surrounds is probably a compound of carbon. Foods, fuels, textiles, drugs everything is either a compound of carbon or its substituent. So let us study about some important

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carbon compounds.

Some Important Carbon Compounds: Types, Videos and Examples

Arrange the following in order of size,
from smallest to largest: polymer,
monomer, carbon atom, macromolecule.
Carbon atom, monomer, polymer,

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macromolecule Explain how a water molecule is produced when glucose and fructose undergo a condensation reaction.

Biology Chapter 3: Section 3-2 Review: Carbon Compounds ...

Questions Answer discussion of carbon and its compound in odia/Chapter-4/Ten

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class Physical science. Hi I am sushanta kumar mohanta Msc(physics) student of centurian university bhubaneswar.I ...

Questions Answer discussion of carbon and its compound in odia/Chapter-4/Ten class Physical science.

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This section explains how the element carbon is able to form millions of carbon, or organic, compounds. It also describes

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the four groups of organic compounds found in living things. The Chemistry of Carbon (page 44)

Macromolecules The Chemistry of Carbon

Study Guide KEY CONCEPT Carbon-based molecules are the foundation of life. VOCABULARY monomer lipid amino

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acid polymer fatty acid nucleic acid
carbohydrate protein MAIN IDEA: Carbon
atoms have unique bonding properties.

1. Why is carbon often called the
building block of life? 2. What ability
allows carbon atoms to form a large
number of molecules? 3.

SECTION ATOMS, IONS, AND

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2.3 Carbon Compounds Lesson

Objectives Describe the unique qualities of carbon. Describe the structures and functions of each of the four groups of macromolecules. Lesson Summary The Chemistry of Carbon Organic chemistry is the study of compounds with bonds between carbon atoms. Carbon atoms

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have four valence electrons, allowing them to form

Chapter 2 The Chemistry Of Life Study Guide Answer Key

Carbon atoms have four valence electrons, allowing them to form strong covalent bonds with many other elements, including hydrogen, oxygen,

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phosphorus, sulfur, and nitrogen. Living organisms are made up of molecules made of carbon and these other elements. One carbon atom can bond to another to form chains and rings.

2_3_study_guide - Name Class Date

2.3 Carbon Compounds ...

1. First, identify the parent chain (

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Answer Key

carbon chain with the most numbers of carbon atoms and also the most prioritize group is present). 2. Identify the substituents. 3.

Draw the IUPAC name of the following compounds | Study.com

Answer to The study of carbon-containing compounds and their

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Answer Key

properties is called organic chemistry.

Solved: The study of carbon-containing compounds and their ...

C₇H₉NO₃ c. C₇H₉NO₂

d. C₁₄H₁₈N₂O₅ | Study.com

Answer to: A compound contains 57.2 percent carbon, 6.1 percent hydrogen, 9.5 percent nitrogen, and 27.2 percent

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oxygen. Which of the...

**A compound contains 57.2 percent
carbon, 6.1 ... - study.com**

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Study Guide - Chapter 3 Elements, Compounds, and Mixtures Section 1: Elements Pages 56-59 ... Circle the letter of the best answer for each question. 5. Why does a helium-filled balloon float up when you let go? ... c. carbon dioxide d. nitrogen compounds 33. What do plants use during photosynthesis to make carbohydrates? a. soil b. carbon ...

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Study Guide Chapter 3 Elements, Compounds, and Mixtures

Carbon atoms form four bonds with no lone pairs, and hydrogen atoms form one bond with no lone pairs. To achieve these bonding patterns, there must be a double bond between the carbon atoms.

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Chapter 5 Chemical Compounds

AP Biology Cell Respiration Quiz Study Guide ... Four of the five answers listed below are degradation processes for carbon compounds. Select the exception. a. Calvin-Benson cycle b. Krebs cycle ... 14. b 15. c 16. C 17. D 18. E 19. E 20. D 21. D 22. A . Title: Microsoft Word - AP Biology Cell

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Respiration Quiz Study Guide
ANSWERS.doc Author ...

AP Biology Cell Respiration Quiz Study Guide ANSWERS

carbon-containing compound, such as benzene, ethanol, glycerol, glucose, fructose, sucrose, ATP, and ADP. 2. A functional group is a cluster of atoms in

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a com-pound that influences the properties of that com-pound. Examples: hydroxyl group, phosphate group. 3. An alcohol is an organic compound with a hydroxyl group attached to one of its carbon atoms.

VOCABULARY REVIEW Define the following terms.

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1. Organic compounds contain a. carbon and usually other elements. b. many kinds of elements except carbon. c. only carbon. d. only carbon and hydrogen.
2. The number of covalent bonds a carbon atom can form with other atoms is 4.
3. A covalent bond formed when two atoms share two pairs of electrons is called a a. -single bond. b. double bond.
- 4.

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